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Chemistry Chapter 9
Chemical Names and
Formulas. STUDY.
PLAY. Monatomic ions.
Consist of a single
atom with a positive or

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Chemical Names

negative charge resulting from the loss or gain of one or more valence electrons, respectively. Key 1.

When the metals in Group 1A, 2A, and 3A lose electrons, they form cations with positive charges equal to their group number.

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Chapter 9 - Chemical

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Gravity. Created by.

bibiloni1901. Terms in
this set (20) Monatomic
Ion. Consists of a single
atom with a positive or
negative charge
resulting from the loss
of gain of one or more
valence electrons and
behaves as a unit, yet
it still has a ...

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Page 6/25

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Chemical Names

Names and Formulas

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29 terms, haless913.

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Names and Formulas.

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periodMs. Steele.

STUDY. PLAY.

Monatomic ion. - single atom with a positive or negative charge resulting from the loss or gain of electrons, respectively.

Chapter 9: Chemical Names and Formulas

Page 7/25

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Chapter 9

Chemical Names And Formulas Practice Problems Answer Key

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Chapter 9 Chemical

names and formulas.

Chemical formula.

Empirical formula.

molecular formula.

functional groups.

notation that shows the
relative # of atoms in a

molecule. a formula

that shows the lowest

whole # ratio of

elements in a.... a

formula that shows the

true ratio of elements

in a compound.

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Chemistry - Chemical Names and Formulas - Chapter 9. Monatomic ions. Polyatomic ions. Binary compound.

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Answer Key

Acid. A single atom with a positive or negative charge. A tightly bound group of atoms that behaves as a unit and has.... A compound composed of two elements that can be either ionic o....

chemistry

vocabulary chapter

9 chemical names

Flashcards ...

Chapter 9 "Chemical Names and Formulas"

H 2 O 2. Section 9.1

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Chapter 9

Chemical Names Naming Ions

OBJECTIVES:

- Identify the charges on monatomic ions by using the periodic table, and name the ions.
-
-

Chemistry - Chp 9 - Chemical Names and Formulas - PowerPoint

Section 9.4 – Naming
and Writing Formulas
for Acids and Bases. An

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Chapter 9

Chemical Names

acid is a compound that produces H^+ ions when it dissolves in water. The formula for an acid normally starts with and H. When naming acids, you should first determine whether or not the acid contains oxygen.

Chapter 9 - Chemical Names and Formulas

Chapter 9 Chemical Names and Formulas 83
SECTION 9.3 NAMING AND WRITING

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Chapter 9

Chemical Names

AND FORMULAS

FORMULAS FOR
MOLECULAR
COMPOUNDS (pages
268-270) This section
explains the rules for

naming and writing
formulas for binary
molecular compounds.

Naming Binary

Molecular Compounds

(pages 268-269)

Name Date Class

CHEMICAL NAMES

AND FORMULAS 9

Naming Ions &

Counting Atoms:

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Chapter 9

Chapter 9.1 Naming &

Writing Formulas for

Ionic Compounds:

Chapter 9.2 Naming &

Writing Formulas for

Molecular Compounds:

Chapter 9.3 Naming &

Writing Formulas for

Acids: Chapter 9.4

Quizlet Chapter 9

vocabulary Answer

Keys Ionic Compound

Naming and Formula

Examples from notes

Chapter 9: Chemical Names & Formulas

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Chapter 9

Chemical Names

Name Date Class

CHEMICAL NAMES AND

FORMULAS 9. Chapter

9 Chemical Names and

Formulas 85 Names and

Formulas for Bases

(page 273) 6. A base is

a compound that

produces _____ when

dissolved in water. 7.

How are bases named?

Chapter 9 Chemical

Names And

Formulas Test B

Answers

Chemical Names and

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Chapter 9

Formulas 281 CHAPTER

9 Assessment 42. a. 2-

b. 1+ c. 1- d. 3+ 43. a.

2+ b. 2+ c. 3+ d. 1+

44. a. barium ion b.

iodide ion c. silver ion

d. mercury(II) ion 45.

cyanide, CN^- and

hydroxide, OH^- 46. a.

hydroxide ion b.

lead(IV) ion c. sulfate

ion d. oxide ion 47.

zero; A compound is

electrically neutral. 48.

The symbols for the

cation and

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Chemical Names
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- Page 303 12 including

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Chemical Names and

Practice Problems

Answer Key

Chapter 9 Chemical Names and

Formulas 211. Section
Review. Objectives.

- Determine the charges of monatomic ions by using the periodic table and write the names of the ions.
- Define a polyatomic ion and write the names and formulas of the most common

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Chemical Names

polyatomic ions.

- Identify the two common endings for the names of most polyatomic ions.

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3:29 PM Page 211

NAMING IONS 9

Chapter 9 Chemical Names and Formulas
221 Name Date 4 Write formulas for the following ionic compounds a sodium phosphate c sodium hydroxide e

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Chapter 9

Chemical Names
ammonium chloride b

And Formulas
magnesium d

Practice Problems
potassium cyanide 5

Answer Key
Write formulas for
compounds formed

from these pairs of ions

b , 6 Name the

following compounds ...

[MOBI] Chapter 9 Chemical Names And Formulas Answers

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Names and Formulas"
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Practice Problems

Answer Key

activities; ... In samples of any chemical compound, the masses of the elements are always in the same proportions any atom or group of atoms that has a negative charge ... place cation name first followed by the anion name: naming polyatomic ...

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Chapter 9

Chemical Names
Formulas Chapter 9

Quiz Chemical Names

And Formulas Chapter

Unit 5: Chemical

Names & Formulas

Chemical Names &

Formulas (Chapter 9) 2

3 Monatomic Ion

Formation &

Nomenclature

Monatomic Ions consist

of a _____ atom with a

positive or negative

charge resulting from

the gain or loss of one

or more valence _____

Cation: any atom or

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Chemical Names

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Quiz**

Chemical Names and
Formulas 281 CHAPTER
9 Assessment 42 a 2- b
1+ c 1- d 3+ 43 a 2+ b
2+ c 3+ d 1+ 44 a
barium ion b iodide ion
c silver ion d
mercury(II) ion 45
cyanide, CN^- and
hydroxide, OH^- 46 a
hydroxide ion b
lead(IV) ion c sulfate

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Chemical Names

ion d oxide ion 47 zero;

And Formulas

electrically neutral 48

The symbols for the

cation and

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